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ance Notes on Codes and Abbreviations" appearing at the begin-  
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(54) Title: CURING AGENTS FOR EPOXY RESINS, USE THEREOF AND EPOXY RESIN CURED THEREWITH

(57) Abstract: Curing agents for epoxy resins are obtained by A) an addition of at least one aldehyde or ketone to at least one phosphonous acid derivative, followed by B) a condensation reaction of the adduct obtained with at least one diamino or polyamino compound. The resulting curing agents can be used alone or as co-curing agents for epoxy resins, and simultaneously as fire-retardants. Using the curing agents of the invention it is possible to prepare cured epoxy resins modified to be fire retardant which have a phosphorus content of at least 2.2 % by weight without lowering the glass transition temperature ( $T_g$ ) to below 150°C. Use of the curing agents and epoxy resins produceable therewith are other objects of the present invention.

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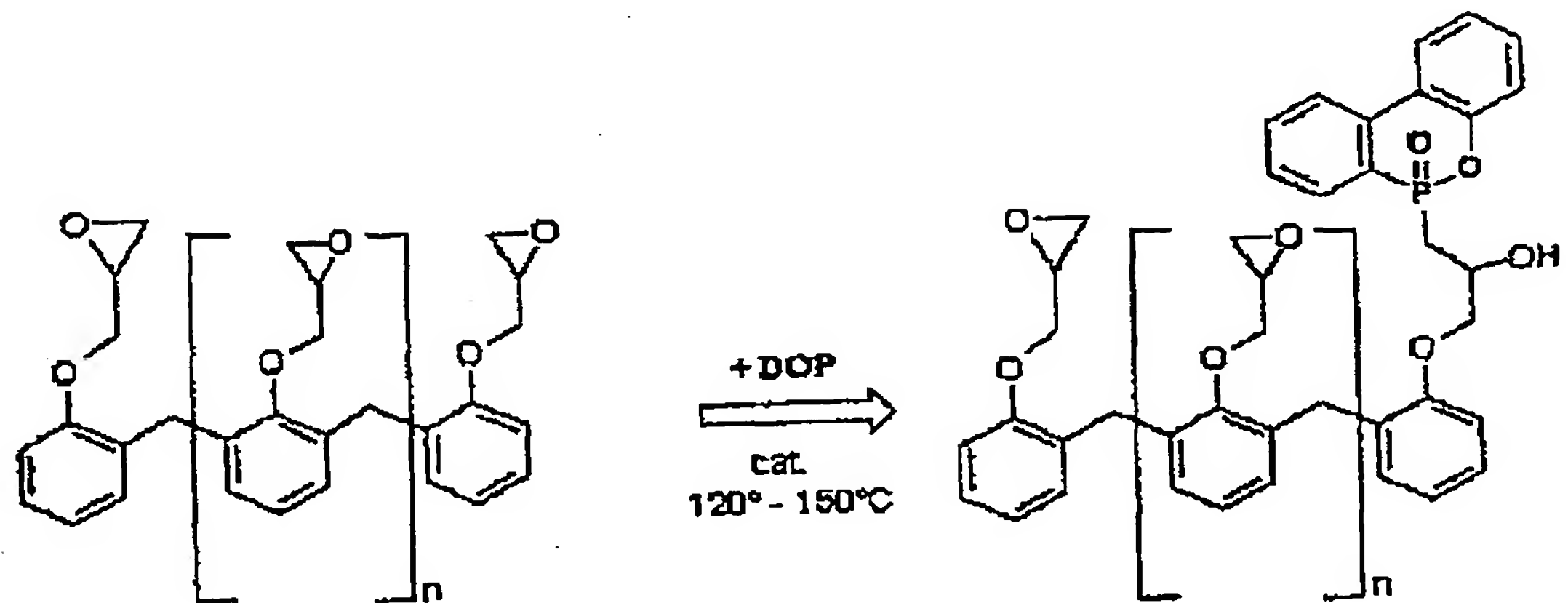
Curing agents for epoxy resins, use thereof and epoxy resin cured therewith

The invention relates to new, phosphorus-containing curing agents for epoxy resins which at the same time provide fire-inhibiting effects.

5       Commercially available epoxy resins are usually liquid, viscose, glassy or also crystalline substances which before being used are mixed with adequate curing agents and optionally also modified with solvents, reactive diluents (viscosity lowering epoxy resins), softeners, fillers, or pigments. As curing agents are used, for example, dicarboxylic acid anhydrides or amine curing agents such as  
10       diethylene triamine. There are systems of epoxy resins and curing agents which at room temperature cure immediately after the epoxy resin and curing agent are brought together, and there are systems which cure only after a certain "working time" after the epoxy resin and curing agent have been brought together, or only after a relatively high temperature has been exceeded. Since in the last mentioned  
15       systems the curing effect first occurs following a certain delay, the curing agents used therefor are occasionally designated as "latent curing agents."

      The process of making epoxy resins fire retardant, as prescribed for many applications, previously was carried out by incorporating traditional flame-protection agents such as antimony trioxide or highly brominated organic  
20       compounds into the epoxy resin compositions. (See, for example, "International Plastics Flammability Handbook" 2nd edition, Carl Hanser Publishing House, Munich, 1990). Aside from the toxicological risks, the use of such traditional flame-protection agents often led to impairment of the mechanical and physical properties of the epoxy resins incorporating such agents therein. Therefore, it  
25       previously has been proposed to use reactive flame-protection agents which are bonded to the polymer structure of the epoxy resin such as, for example 10-Oxo-10H-9-oxa-10-phospha-phenanthrene known as "DOP" which can also be designated as 9,10-Dihydro-9-oxa-10-phospha-phenanthrene-10-oxide (see EP-B1-0 806 429, Claim 24).

In a trifunctional epoxy-Novolac resin, DOP reacts with the glycidyl groups of the resin according to the following reaction diagram:



5 However, the above-described process for making the epoxy resins fire retardant with DOP is insufficient for many applications for the following reasons:

- For standard fire-protection modifying of epoxy resins there is required a phosphorus content of more than 2.2% by weight based on the total mass of the modified and cured epoxy resin. To attain such phosphorus content the portion of DOP had to comprise at least 60% of the modified resin.

- Such a high portion of fire-protection agent results in a reduction of the glass transition temperature ( $T_g$ ). Depending on the epoxy resin, there can thus be used only glass transition temperatures of less than 150°C, dynamically measured by Dynamic Scanning Calometry ("DSC"). A  $T_g$  typical for such systems is 135°C (DSC). But epoxy resins to be used for preparing printed circuit board substrates or for other electronic purposes such as casting resins or embedding masses must have, with regard to the lead-free tin solder compositions in use at present, having melting points between about 185 and 205°C, a glass transition temperature of at least 150°C, preferably more than 160 or 165°C (DSC).

- DOP can only be used for a few epoxy resins as fire-protection agent directly bonding to the glycidyl groups. However, in case of a slight change of the chemical structure of the resin, for example, changing from phenol-Novolac resin to cresol-Novolac, incompatibility problems can appear.

5 DE-OS 100 06 592 discloses "latent" combination compounds consisting of amine curing agents for epoxy resins and fire-protection agents as, for example, DOP, the same as the use thereof for preparing cured, fire-retardant modified epoxy resins which, when present as one-component system, exhibit a working  
10 time of several hours during which no curing occurs, or which cure only when a certain relatively high threshold temperature is exceeded. In these known latent curing agents the nitrogen atom of the amino groups of an amine curing agent is directly bonded to the phosphorus atom of DOP.

The problem which the present invention addresses is to provide novel curing agents for epoxy resins which simultaneously exhibit curing and flame-inhibiting  
15 characteristics, which either alone or as co-curing agents have a phosphorus content required for the desired fire-protection of at least 2.2% by weight of the cured epoxy resin without lowering the glass transition temperature  $T_g$  of the resin to a value below  $150^{\circ}\text{C}$  (DSC).

According to the invention this problem is solved by a curing agent for epoxy  
20 resins that can be obtained by the steps of:

(A) An addition of at least one aldehyde or ketone of the formula I:



wherein  $\text{R}^1$  and  $\text{R}^2$  independently of each other are selected from a hydrogen atom, an optionally substituted  $\text{C}_1 - \text{C}_8$ -alkyl, aryl, alkaryl, or aralkyl group, to at  
25 least one phosphonous acid derivative of the formula II:



wherein  $R^3$  and  $R^4$  independently of each other are selected from an optionally substituted  $C_1 - C_8$ -alkyl, cycloalkyl, aryl, alkaryl, aralkyl, or heteroaryl group, or  $R^3$  and  $R^4$  together are selected from a mononuclear or polynuclear, optionally substituted, aromatic or non-aromatic ring system (first step) and

(B) a condensation reaction of the adduct obtained in the first step with at least one diamino or polyamino compound (second step).

It surprisingly has been found that when using the curing agent of the present invention for curing and making fire-retarding epoxy resins, the required phosphorus content for standard flame-protection modification of at least 2.2% by weight can be reached without harmful reduction of the glass transition temperature of the resin to below 150°C. There was on the contrary found an increase of the glass transition temperature to up to 170°C (DSC) with a phosphorus content of the resin of about 3% by weight. Thereby can be satisfied the need of halogen-free, fire-retardant and cured epoxy resin systems which can be used particularly for the preparation of printed circuit board substrates and other electronic parts which come into contact with lead-free solder tin of high melting point.

For the curing agents of the present invention, the aldehydes preferably  
comprise formaldehyde, paraformaldehyde, a formaldehyde rendering compound  
such as 1,3,5-trioxane or an optionally substituted benzaldehyde while as ketones  
can preferably be used an optionally substituted acetophenone or benzophenone.

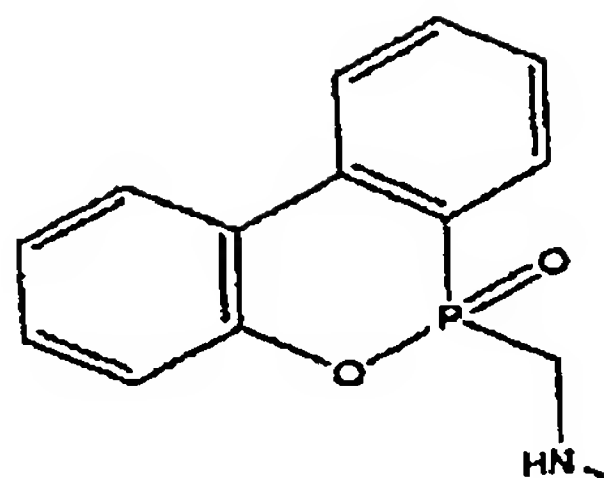
As diamino or polyamino compounds preferably ethylene diamine, 4,4'-diamino-diphenyl methane, 4,4'-diamino-diphenyl sulphone (DDS), urea, or  
25 melamine can be used.

The addition reaction in the first step of preparation of the curing agents of the present invention can preferably be carried out at a temperature of from 50 to 150°C, while the condensation reaction in the second step can be carried out preferably at a temperature of from 100 to 200°C, optionally in the presence of a catalyst.

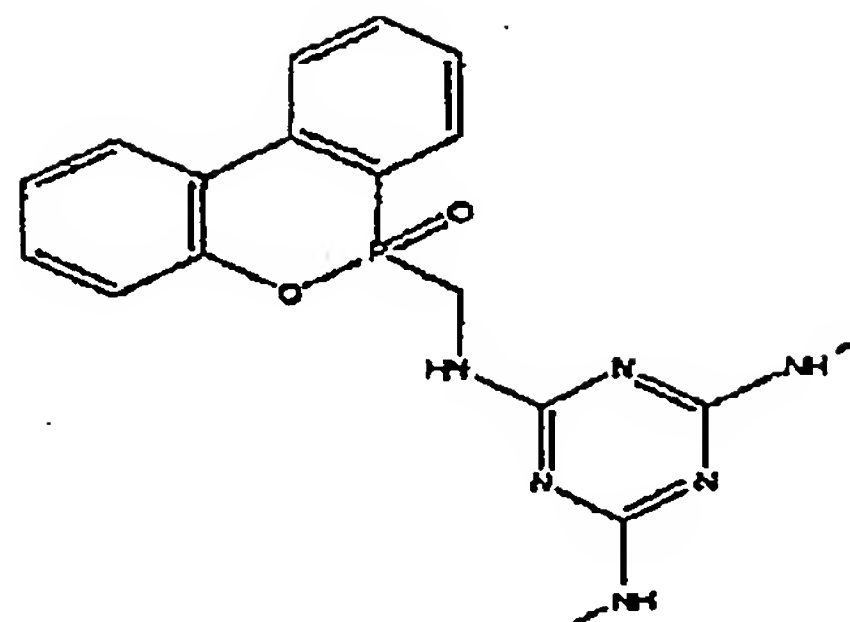
For the preparation of the curing agent of the present invention, as phosphonous acid derivative there is preferably used the compound 10-Oxo-10H-9-oxa-10-phospha-phenanthrene (equivalent to 9,10-Dihydro-9-oxa-10-phospha-phenanthrene-10-oxide) known as "DOP" or a mixture containing DOP and 2'-hydroxydiphenyl-2-phosphinic acid, as known, for example from EP-B1-0833832.

In the preparation of the curing agent of the present invention, the molar ratio of the components (a) aldehyde or ketone, (b) phosphonous acid derivative and (c) diamino or polyamino compound is preferably from  $a:b:c=1:1:1$  to  $a:b:c=x:x:1$  wherein  $x$  corresponds to the number of amino groups in one molecule of the polyamino compound.

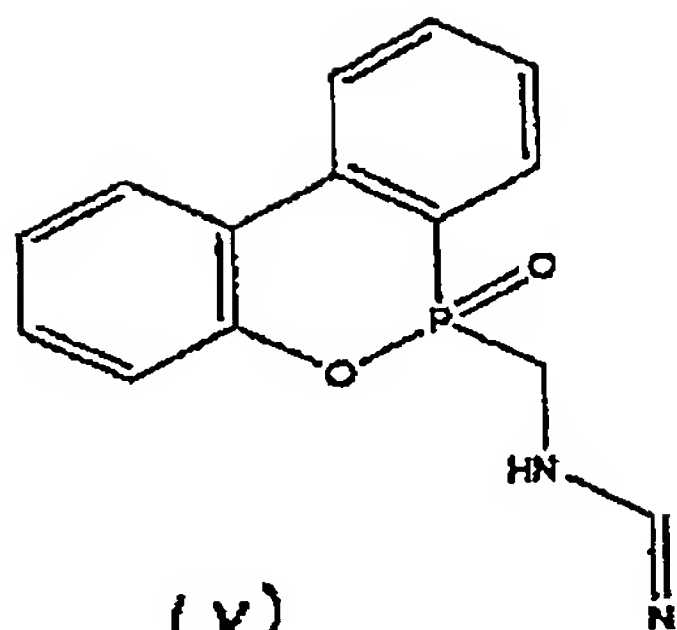
Preferred embodiments of the inventive curing agent are characterized by a molecular structure containing at least one radical of the formulae III, IV, V or VI:



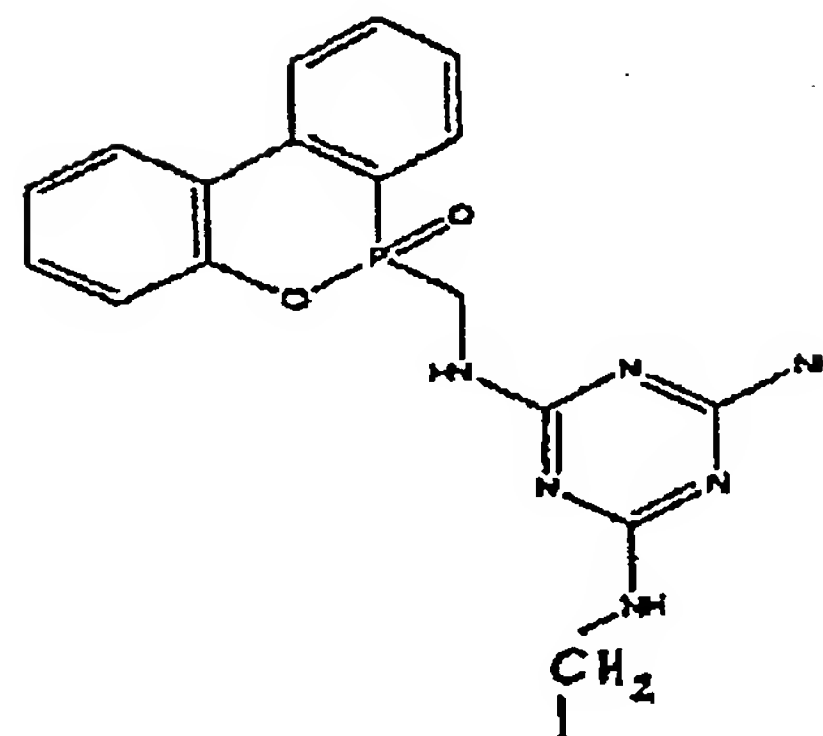
( III )



( IV )



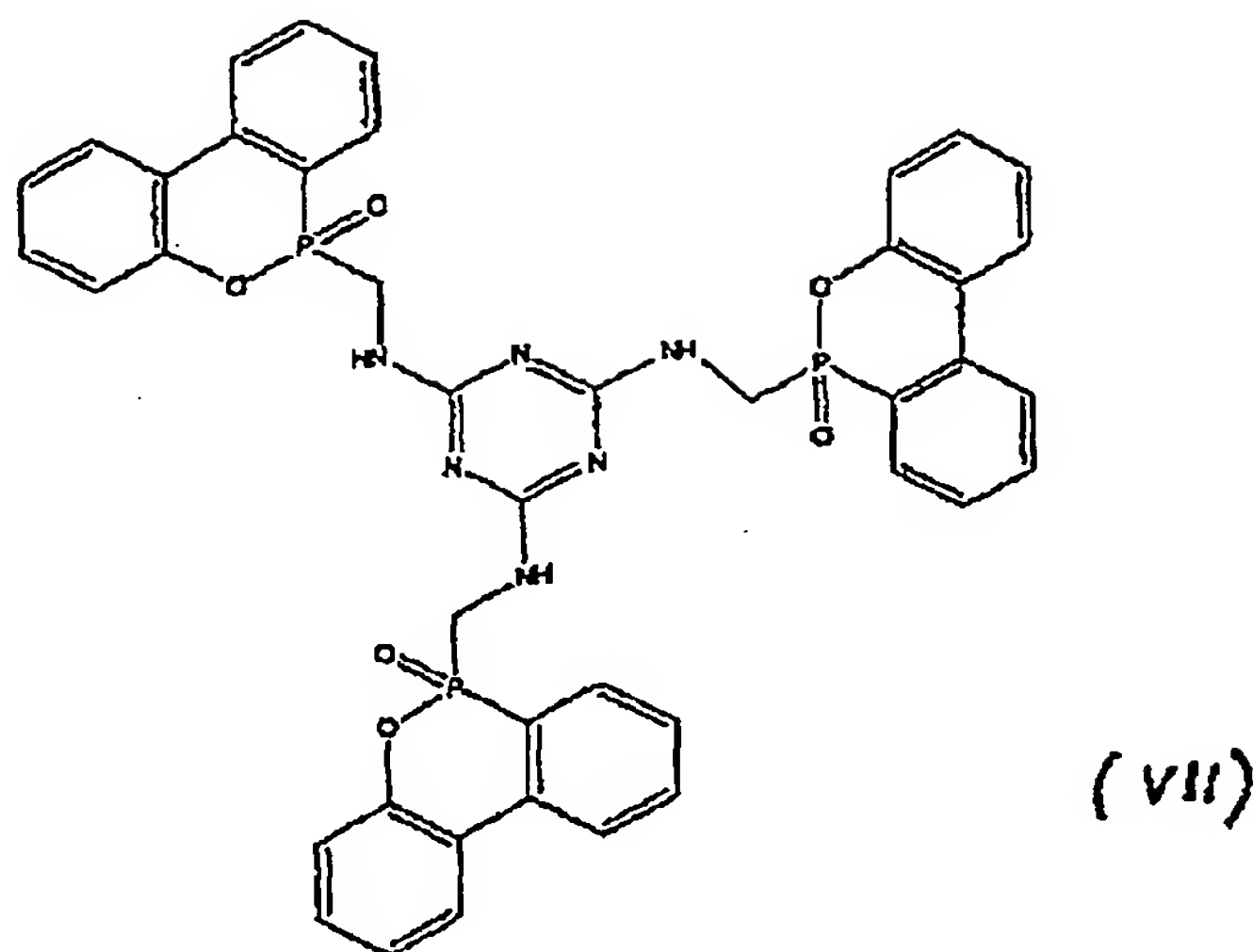
( V )



( VI )

One specially preferred curing agent of the present invention is N,N',N''-Tris-(10-oxo-10H-9-oxa-10-phospha-phenanthrene-10-ylmethyl) -[1,3,5] triazine-2, 4, 6-triamine which has the structural formula VII:





The curing agents of the present invention can be used as a single curing agent or as co-curing agent for epoxy resins, which means together with at least one other curing agent for epoxy resins, preferably together with an amine curing agent, specially one or more of the group consisting of diethylene triamine, dimethyl amino propylamine, isophoron-diamine, dicyan diamide (cyanoguanidine).

Specially advantageous is the use of the curing agent of the present invention for raising the glass transition temperature of the cured epoxy resin to above 150°C, particularly to above 165°C (DSC in each case).

Likewise particularly advantageous is the use of the curing agent of the present invention for flameproofing an epoxy resin, preferably preferred for preparing cured epoxy resins modified to be fire retardant and having a phosphorus content of at least 2.2% by weight, preferably of 2.8 to 3.2% by weight.

Another object of the present invention is constituted by epoxy resins cured and modified to be fire retardant with a curing agent of the present invention, said epoxy resins being preferably epoxy-Novolac resins.



The epoxy resin cured and modified to be fire retardant according to the present invention serves preferably for making printed circuit board substrates and other products in the field of microelectronics in which, due to the contact with molten lead-free tin solder usually having a melting point between about 185 and  
5 205°C, a high glass transition temperature of at least 150°C (DSC) and at the same time an effective flame protection are especially important.

The invention is explained in detail herebelow with reference to the following working example:

### Example

#### 10 Preparation of the curing agent

##### A) First step

40g (185 mmol) DOP were heated to 130°C, 15 g formaline solution (37% by weight) were added drop by drop at this temperature over 45 minutes while stirring. After another 30 minutes a white solid body formed which began to melt  
15 at about 140°C.

##### B) Second step

The adduct generated in the first step was further heated to about 150°C. To the resulting clear solution were added 7.8 g melamine (61.7 mmol) and the mixture was heated over two hours to 200°C. The mixture was kept at 200°C until  
20 no more water was formed.

#### Curing and flameproofing an epoxy resin

7 g of an epoxy-Novolac resin (LER N 740 available from the firm LG, Korea; epoxide number: 0.575 mol/100g) and 3g of an epoxy-Novolac resin (Ukanol EPUK available from the firm Schill + Seilacher, Boeblingen; epoxide number:  
25 0.28 mole/100 g) were heated with 3 g of the curing agent of the present invention and prepared as above to 160°C and stirred until a clear solution formed. Into this

solution was incorporated 2.4 g diamino diphenyl sulphone in solid form and stirred until a clear solution formed. The solution was held for 15 minutes at 160°C, then 30 minutes at 175°C and finally 3 hours at 195°C and thereby cured. There resulted a yellow transparent solid body of cured epoxy resin  
5 simultaneously modified to be fire retardant. The glass transition temperature of the resin prepared was 170°C (DSC) and had a phosphorus content of 2.9% by weight. The resulting cured sample satisfied the UL94 V-0 fire protection standard of Underwriters' Laboratories (USA).

Claims

1. A curing agent for epoxy resins obtainable by

A) the addition of at least one aldehyde or ketone of the formula I:



5 wherein  $\text{R}^1$  and  $\text{R}^2$  independently of each other designate a hydrogen atom, an optionally substituted  $\text{C}_1 - \text{C}_8$ -alkyl, aryl, alkaryl, or aralkyl, to at least one phosphonous acid derivative of the formula II



10 wherein  $\text{R}^3$  and  $\text{R}^4$  independently of each other designate an optionally substituted  $\text{C}_1 - \text{C}_8$ -alkyl, cycloalkyl, aryl, alkaryl, aralkyl, heteroalkyl, or  $\text{R}^3$  and  $\text{R}^4$  together are selected from a mononuclear or polynuclear, optionally substituted, aromatic or non-aromatic ring system (first step); and

B) a subsequent condensation of the adduct obtained in the first step with at least one diamino or polyamino compound (second step).

15 2. The curing agent of claim 1, characterized in that formaldehyde, paraformaldehyde, a formaldehyde rendering compound such as 1,3,5-trioxane or an optionally substituted benzaldehyde is used as said aldehyde.

3. The curing agent of Claim 1, characterized in that an optionally substituted acetophenone or benzophenone is used as said ketone.

20 4. The curing agent of any one of claims 1 to 3, characterized in that the addition reaction in the first step is carried out at a temperature of from 50 to 150°C.

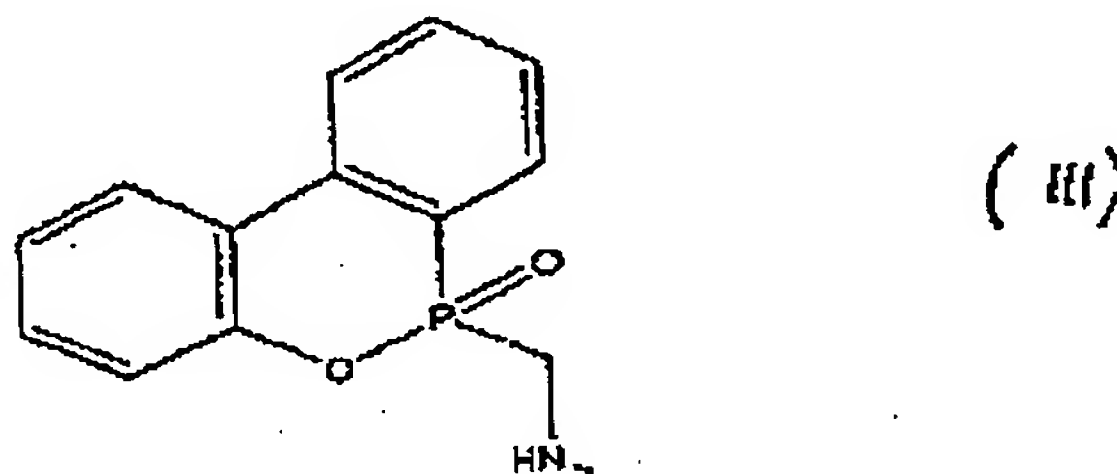
5. The curing agent of any one of claims 1 to 4, characterized in that ethylene diamine, 4,4'-diamino-diphenyl methane, 4,4'-diamino-diphenyl sulphone (DDS), urea or melamine is used as said diamino or polyamino compound.

6. The curing agent of any one of claims 1 to 5, characterized in that the  
5 condensation reaction in the second step is carried out at a temperature of from 100 to 200°C, optionally in the presence of a catalyst.

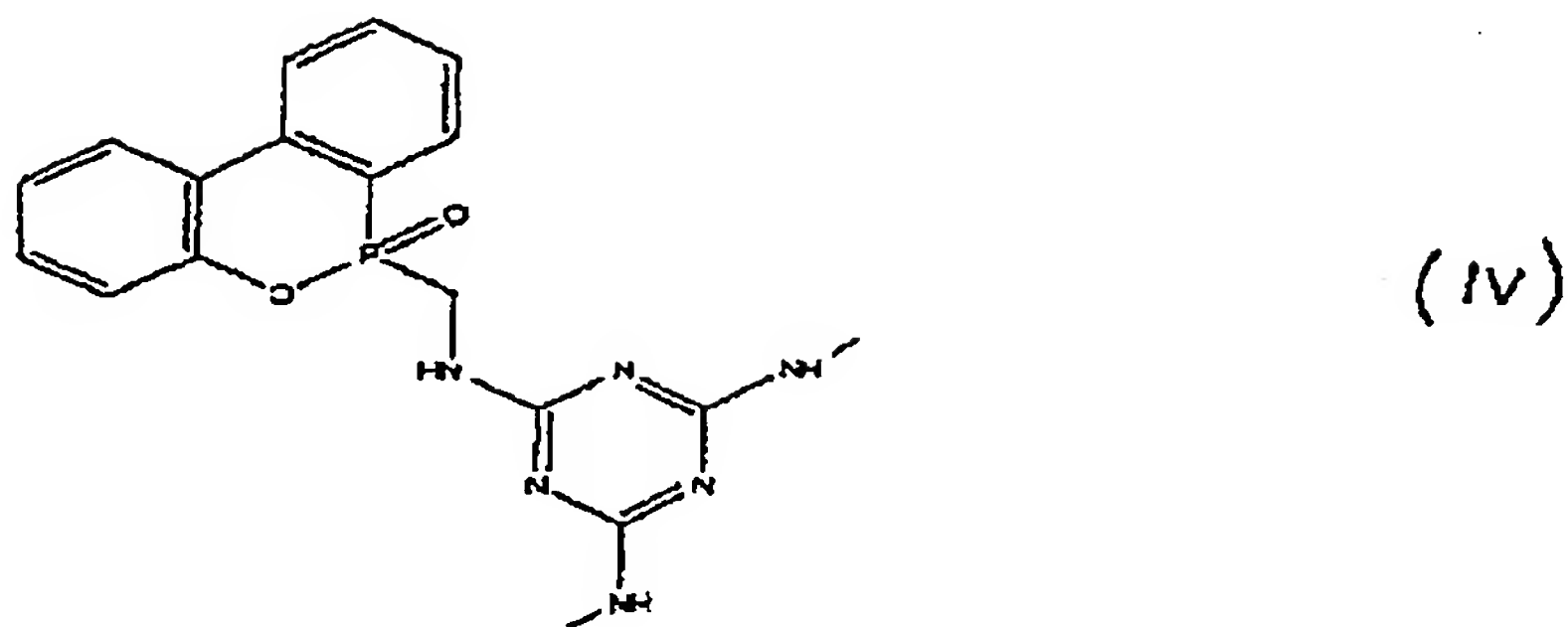
7. The curing agent of any one of claims 1 to 6, characterized in that 10-Oxo-  
10H-9-oxa-10-phospha-phenanthrene (9,10-Dihydro-9-oxa-10-phospha-phenan-  
threne-10-oxide) ("DOP") or a mixture containing DOP and 2'-hydroxydiphenyl-2  
10 -phosphinic acid is used as said phosphonous acid derivative.

8. The curing agent of any one of claims 1 to 7, characterized in that it is  
obtainable by reacting the components (a) aldehyde or ketone, (b) phosphonous  
acid derivative and (c) diamino or polyamino compound in a mole ratio of from  
a:b:c = 1:1:1 to a:b:c = x:x:1, wherein x corresponds to the number of amino  
15 groups in one molecule of the polyamino compound.

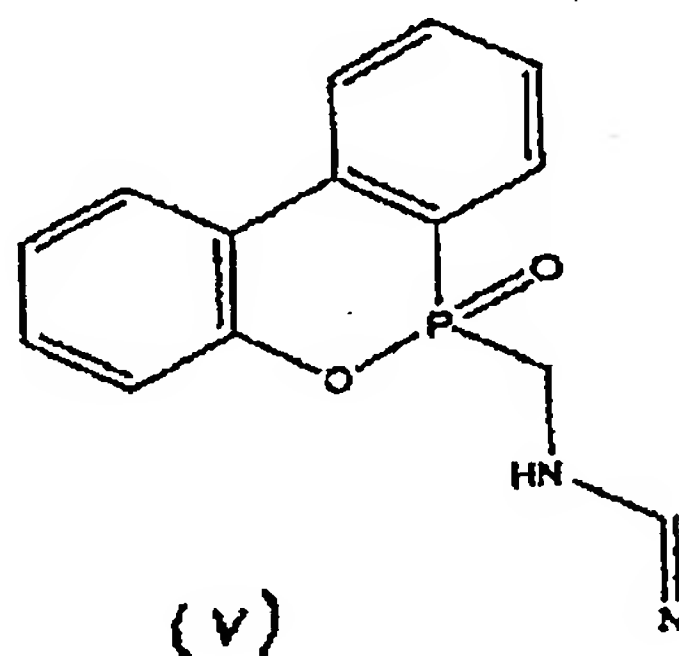
9. The curing agent of any one of claims 1 to 8, characterized by a molecular  
structure containing at least one radical of the formula III:



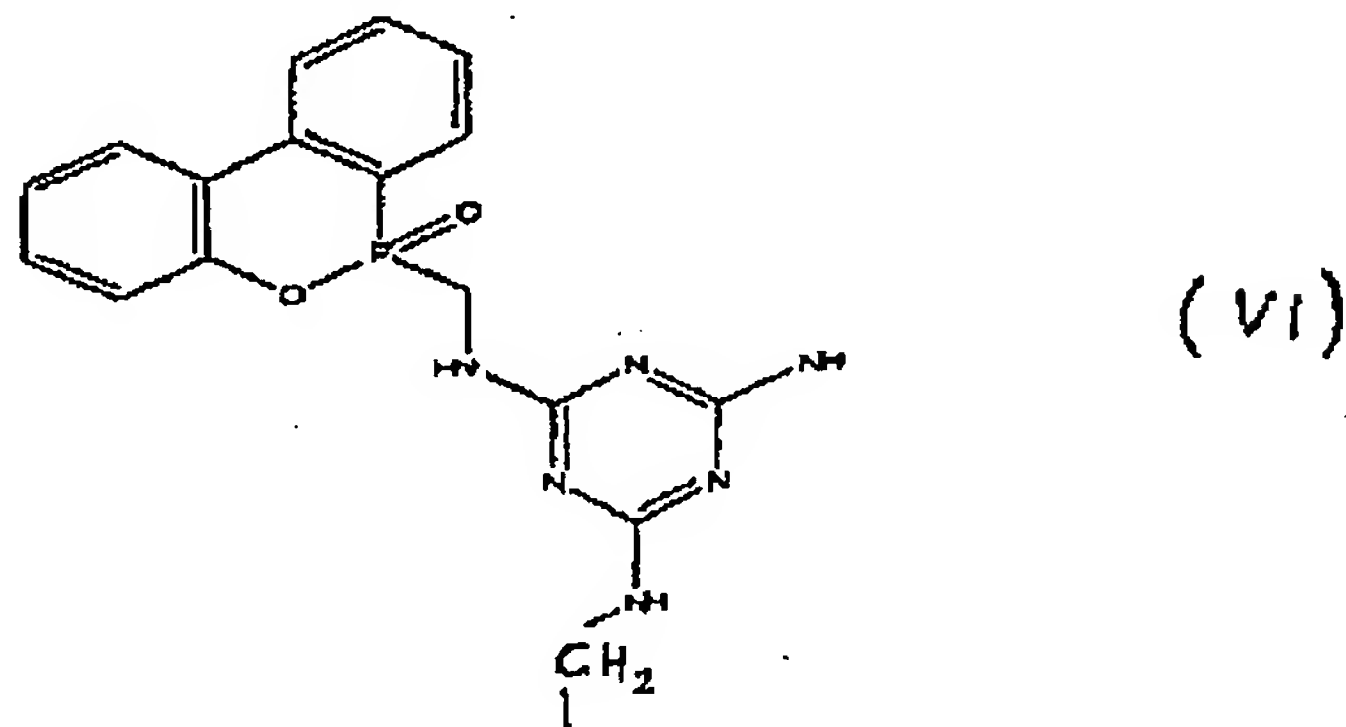
10. The curing agent of claim 9, characterized by a molecular structure  
20 containing at least one radical of the formula IV:



11. The curing agent of claim 9, characterized by a molecular structure containing at least one radical of the formula V:



5 12. The curing agent of claim 10, characterized by a molecular structure containing at least one radical of the formula VI:



13. The curing agent of claim 12, namely N,N'N''-Tris-(10-oxo-10H-9-oxa-10-phospha-phenanthrene-10-ylmethyl)-[1,3,5] triazine-2,4,6-triamine.

14. Use of the curing agent of any one of claims 1 to 13 as a co-curing agent for epoxy resins, together with at least one other curing agent for epoxy resins.

15. The use of claim 14, together with an amine curing agent, especially with one or more of the group consisting of diethylene triamine, dimethyl aminopropylamine, isophorondiamine and dicyan diamide (cyanoguanidine).

16. The use of claim 14 or 15 for raising the glass transition temperature ( $T_g$ ) of a cured epoxy resin to above 150°C (DSC).

17. The use of claim 16 for raising the glass transition temperature of a cured epoxy resin to above 165°C (DSC).

18. The use of any one of claims 14 to 17 for flameproofing an epoxy resin.

19. The use of claim 18 for producing cured epoxy resins modified to be fire retardant and having a phosphorus content of at least 2.2% by weight.

20. The use of claim 19 for producing cured epoxy resins modified to be fire retardant and having a phosphorus content of from 2.8 to 3.2% by weight.

21. An epoxy resin cured and modified to be fire retardant with a curing agent according to any one of claims 1 to 13.

22. The epoxy resin of claim 21, characterized in that it is an epoxy-Novolac resin.

23. The epoxy resin of claim 21 or 22, characterized in that it is a printed circuit board substrate.

# INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 03/10067

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C08G59/18 C08G59/68 C07F9/30

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C08G C07F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the International search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 6 441 067 B1 (CHIU YIE-SHUN ET AL) 27 August 2002 (2002-08-27) the whole document	1-23
A	DE 11 54 624 B (SCHERING AG) 19 September 1963 (1963-09-19) the whole document	1-23
A	US 2001/014706 A1 (DOERING MANFRED ET AL) 16 August 2001 (2001-08-16) the whole document	1-23

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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Date of the actual completion of the international search

3 November 2003

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